

## PHENYLPROPANOID GLYCOSIDES FROM *CALCEOLARIA HYPERICINA*\*

M. NICOLETTI, C. GALEFFI†, I. MESSANA, G. B. MARINI-BETTOLO, J. A. GARBARINO‡, and V. GAMBARO‡

Dipartimento di Biologia Vegetale, La Sapienza, P. le A. Moro, 2, and Centro CNR di Chimica dei Recettori e delle molecole biologicamente attive, Istituto di Chimica, Università Cattolica del S. Cuore, Roma, Italy; †Laboratorio di Chimica del Farmaco, Istituto Superiore di Sanità, Roma, Italy; ‡Universidad Federico Santa María, Casilla 110-V, Valparaíso, Chile

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**Key Word Index**—*Calceolaria hypericina*; Scrophulariaceae; phenylpropanoid glycosides; calceolarioside C and D, verbascoside.

**Abstract**—Besides the known compound verbascoside, two new phenylpropanoid glycosides, calceolarioside C, 1'-O-β-D-(3,4-dihydroxy-β-phenyl)-ethyl-4'-O-caffeoxy-β-D-xylopyranosyl-(1''→6')-glucopyranoside, and calceolarioside D, 1'-O-β-D-(1-hydroxy-4-oxo-2,5-cyclohexadienyl)-ethyl-6'-O-caffeoxyglucopyranoside, were isolated from the aerial parts of *Calceolaria hypericina*. The structures of the new compounds were elucidated by spectroscopic methods.

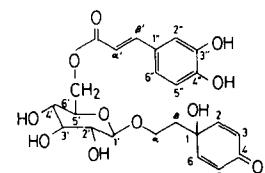
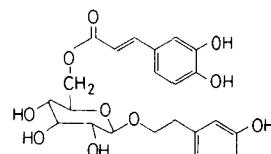
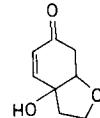
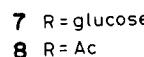
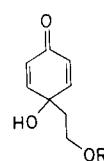
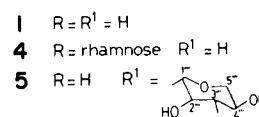
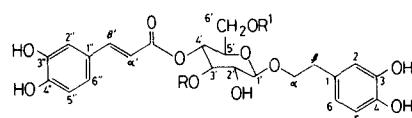
Several species of *Calceolaria* (Scrophulariaceae) are used in popular Chilean medicine as a tonic, stomachic and cicatrizing agents and against bacterial infections [2, 3]. A previous paper described the isolation of two new phenylpropanoid glucosides, calceolarioside A (1) and calceolarioside B (2), from the methanolic extract of *C. hypericina*, besides halleridone (3) [1]. Further purification by counter-current distribution (CCD) of the same extract resulted in the isolation of three additional minor compounds. One of them was readily identified as verbascoside (acteoside) (4) [4] whereas the other two were new phenylpropanoid glycosides, 5 and 6, named calceolarioside C and D, respectively.

IR and UV spectra of calceolarioside C (5), mp 123–125°,  $C_{28}H_{34}O_{15}$ , were very similar to those of calceolarioside A and B [1]. Also  $^1H$  NMR and  $^{13}C$  NMR data (Tables 1 and 2, respectively) were closely related to those of 1, except for the presence of additional peaks attributable to a sugar unit. The deshielding of H-6<sub>a</sub> and H-6<sub>b</sub> ( $\Delta\delta = 0.2$  ppm) allowed assignment of the linkage of the additional unit at position 6' of the glucose, as was also confirmed by the corresponding shift of the C-6' signal in the  $^{13}C$  NMR spectrum ( $\Delta\delta = +5$  ppm), due to the *O*-alkyl substitution.

Total acid hydrolysis of 5 yielded D-glucose and D-xylose.  $^1H$  NMR data confirmed the identification of the additional sugar unit as xylose, linked in the  $\beta$ -form [H-1'' at  $\delta 4.25$ , doublet with  $J = 7.5$  Hz and C(1'') at  $\delta 105.2$  [5], and thus the structure of 1'-O-β-D-(3,4-dihydroxy-β-phenyl)-ethyl-4'-O-caffeoxy-β-D-xylopyranosyl-(1''→6')-glucopyranoside was assigned to calceolarioside C.

Compound 5 therefore is an isomer of conandrioside, isolated from *Conandron ramoidioides* [6], in which the

xylose linkage was assigned to position 3'. A diagnostic difference between the two compounds is represented by the chemical shift value of the anomeric proton of the xylose ( $\delta 4.52$  in conandrioside). As a confirmation in the couple verbascoside (rhamnose in 3') [4]/forsythoside A



\* Part 2 in the series 'Studies in *Calceolaria* genus'. For Part 1 see ref. [1].

Table 1.  $^1\text{H}$  NMR spectral data of compounds **5** and **6**\*

H	<b>5</b>	<b>6</b>
2	6.68 ( <i>d</i> , <i>J</i> = 2.5)	6.97 ( <i>d</i> , <i>J</i> = 10.0)
3	—	6.09 ( <i>d</i> , <i>J</i> = 10.0)
5	6.65 ( <i>d</i> , <i>J</i> = 8.0)	6.09 ( <i>d</i> , <i>J</i> = 10.0)
6	6.53 ( <i>dd</i> , <i>J</i> = 2.5 and 8.0)	6.97 ( <i>d</i> , <i>J</i> = 10.0)
$\alpha_1$	4.03 ( <i>m</i> )	3.94 ( <i>m</i> )
$\alpha_2$	3.72 ( <i>m</i> )	3.66 ( <i>m</i> )
$2\beta$	2.80 ( <i>t</i> , <i>J</i> = 7.5)	2.05 ( <i>m</i> )
1'	4.36 ( <i>d</i> , <i>J</i> = 8.0)	4.28 ( <i>d</i> , <i>J</i> = 8.0)
2'	3.60 ( <i>m</i> )	3.20 ( <i>dd</i> , <i>J</i> = 8.0 and 9.0)
3'	3.62 ( <i>t</i> , <i>J</i> = 9.0)	3.33–3.58
4'	4.90 ( <i>t</i> , <i>J</i> = 9.0)	3.33–3.58
5'	3.72 ( <i>m</i> )	3.33–3.58
6'_a	3.86 ( <i>dd</i> , <i>J</i> = 2.0 and 11.5)	4.33 ( <i>dd</i> , <i>J</i> = 2.0 and 11.5)
6'_b	3.82 ( <i>dd</i> , <i>J</i> = 5.0 and 11.5)	4.49 ( <i>dd</i> , <i>J</i> = 5.0 and 11.5)
2''	7.02 ( <i>d</i> , <i>J</i> = 2.5)	7.08 ( <i>d</i> , <i>J</i> = 2.5)
5''	6.78 ( <i>d</i> , <i>J</i> = 8.0)	6.80 ( <i>d</i> , <i>J</i> = 8.0)
6''	6.92 ( <i>dd</i> , <i>J</i> = 2.5 and 8.0)	6.95 ( <i>dd</i> , <i>J</i> = 2.0 and 8.0)
$\alpha'$	7.56 ( <i>d</i> , <i>J</i> = 15.5)	7.59 ( <i>d</i> , <i>J</i> = 15.5)
$\beta'$	6.26 ( <i>d</i> , <i>J</i> = 15.5)	6.30 ( <i>d</i> , <i>J</i> = 15.5)
1'''	4.25 ( <i>d</i> , <i>J</i> = 7.5)	
2'''	3.15 ( <i>dd</i> , <i>J</i> = 7.5 and 9.0)	
3'''	3.30 ( <i>m</i> )	
4'''	3.37 ( <i>m</i> )	
5'_a	3.21 ( <i>dd</i> , <i>J</i> = 7.5 and 8.0)	
5'_b	3.46 ( <i>m</i> )	

\*400 MHz, in  $\text{CD}_3\text{OD}$  with TMS as internal reference. The values of the coupling constants are in Hz.

(rhamnose in **6'**) [7] the resonance of the anomeric proton of the rhamnose shows a similar difference.

Calceolarioside D(**6**),  $\text{C}_{23}\text{H}_{26}\text{O}_{11}$ , showed UV maximum absorptions at 331 and 290 nm and IR bands at 3400 (*br*), 1700 and 1670  $\text{cm}^{-1}$ . In respect to calceolarioside B(**2**) the  $^1\text{H}$  NMR spectrum of **6** (Table 1) presented analogous signals for the *trans*-caffeyl and the 1,6-disubstituted glucose moieties, whereas the remaining resonances did not agree with the 3,4-dihydroxy- $\beta$ -phenylethoxy moiety present in the phenylpropanoids so far isolated from *Calceolaria*. Indeed this last pattern of peaks was attributed to a cyclohexa-2,5-dienone structure ( $\delta$  6.09 and 6.97, two doublets with *J* = 10 Hz, each accounting for two protons) and a  $\text{CH}_2\text{CH}_2\text{O}-$  sequence (Table 1). These assignments, as well as those of the  $^{13}\text{C}$  NMR spectrum of **6**, were in good accordance with those of structurally related products, i.e. cornoside (**7**) [8] and hallerone (**8**) [9], accounting for the presence of a (1-hydroxycyclohexa-2,5-dien-4-one)-ethoxy unit linked to the position 1' of the glucose. Thus the structure of 1'-*O*- $\beta$ -D-(1-hydroxy-4-oxo-2,5-cyclohexadienyl)-ethyl-6'-*O*-caffeylglucopyranoside was assigned to **6**.

The occurrence in plants of different families of phenylpropanoid glycosides and cyclohexanols, as hallerone (**8**) and halleridone (**3**), suggested a common metabolic pathway [9, 10]. This hypothesis is now endorsed by the presence of the cyclohexa-2,5-dienone structure in calceolarioside D.

Table 2.  $^{13}\text{C}$  NMR spectral data of compounds **4** and **5**\*

C	<b>5</b>	<b>6</b>
1	131.5	68.7
2	117.1 <sup>a</sup>	153.6
3	145.7	127.6
4	144.5	187.3
5	115.3 <sup>b</sup>	127.6
6	121.3	153.6
$\alpha$	72.5	65.6
$\beta$	36.5	40.4
1'	104.3	103.7
2'	75.2 <sup>c</sup>	74.2 <sup>a</sup>
3'	75.8 <sup>c</sup>	77.3
4'	72.5	71.0
5'	74.9 <sup>c</sup>	74.8 <sup>a</sup>
6'	68.5	64.2
1''	127.7	127.2
2''	116.4 <sup>a</sup>	116.1
3''	146.4	146.0
4''	149.6	148.8
5''	114.7 <sup>b</sup>	114.5 <sup>b</sup>
6''	123.0	122.6
$\alpha'$	116.3 <sup>a</sup>	114.9 <sup>b</sup>
$\beta'$	147.7	146.8
COO	168.5	168.7
1'''	105.2	
2'''	74.8 <sup>c</sup>	
3'''	77.5	
4'''	71.1	
5'''	66.8	

\*In  $\text{CD}_3\text{OD}$ ; TMS as internal reference.

<sup>a–c</sup>These values may be interchanged in the same column.

## EXPERIMENTAL

$^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectra were registered on a Bruker AM 400 spectrometer. Separations were performed by CCD with a Craig Post apparatus (200 stages, 10:10 ml, upper and lower phase).

**Separation.** Acteoside (**4**) (0.2 g,  $K_r$  = 0.26), and a mixture of **5** and **6** were obtained from the butanolic residue (1.5 g) by CCD using the solvent system  $\text{H}_2\text{O}$ –AcOEt–*n*-BuOH (10:8:2). The mixture, further purified using  $\text{H}_2\text{O}$ –AcOEt–*n*-BuOH (10:8.5:1.5) gave pure calceolarioside D (0.31 g,  $K_r$  = 1.49) and calceolarioside C (0.44 g,  $K_r$  = 1.11). Acteoside was identified by direct comparison with an authentic sample.

**Calceolarioside C (**5**).** Crystals from AcOEt and *n*-hexane, mp 123–125°.  $[\alpha]_D^{25} = -2.7^\circ$  ( $\text{MeOH}$ ; *c* 1); UV ( $\text{MeOH}$ ),  $\lambda_{\text{max}}$ , nm ( $\log \epsilon$ ): 329, 296, 219 (4.08, 4.00, 4.89); IR (KBr),  $\nu_{\text{max}}$ : 3350, 1690, 1650 and 1060  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR: Tables 1 and 2, respectively. (Found C, 54.97; H, 5.70; calcd for  $\text{C}_{28}\text{H}_{34}\text{O}_{15}$  C, 55.08; H, 5.61%).

**Hydrolysis of calceolarioside C (**5**).** Compound **5** (100 mg) was treated with 1 N  $\text{H}_2\text{SO}_4$  (30 ml) at 100° for 1 hr. The reaction mixture was neutralized with  $\text{BaCO}_3$ , the insoluble material was removed by filtration and the soln extracted with AcOEt. In the aq. soln D-glucose and D-xylose were identified by TLC and through the corresponding  $\beta$ -acetyl derivatives separated by

CCD ( $\text{H}_2\text{O}-\text{Me}_2\text{CO}-\text{cyclohexane}$ , 4:6:7) and compared with authentic specimens.

*Calceolarioside D* (**6**). Colourless amorphous powder.  $[\alpha]_D^{25} = -21.5^\circ$  (MeOH;  $c$  2); UV (MeOH),  $\lambda_{\text{max}}$ , nm (log  $\epsilon$ ): 331 (4.13), 290 (4.05); IR (KBr),  $\nu_{\text{max}}$ : 3400 (br), 2920, 1700, 1670, 1630, 1050  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR  $\delta$ : Tables 1 and 2, respectively. (Found C, 57.59; H, 5.55; calcd for  $\text{C}_{23}\text{H}_{26}\text{O}_{11}$ , C, 57.74; H, 5.48%).

## REFERENCES

1. Nicoletti, M., Galeffi, C., Messana, I., Garbarino, J. A., Gambaro, V., Nyandat, E. and Marini-Bettolo, G. B. (1986) *Gazz. Chim. Ital.* **116**, 431.
2. Navas, L. E. (1979) *Flora de la Cuenca de Santiago de Chile (Scrophulariaceae)* Tomo III, p. 105. Ediciones de la Universidad de Chile, Santiago de Chile.
3. Muñoz, M., Barrera, E. and Meza, I. (1981) *El Uso Medicinal y Alimenticio de Plantas Nativas y Naturalizadas en Chile* n. 23, p. 48. Museo Nacional de Historia Natural, Santiago de Chile.
4. Andary, C., Wylde, R., Laffite, C., Privat, G. and Winternitz, F. (1982) *Phytochemistry* **21**, 1123.
5. Andary, C., Privat, G., Wylde, R. and Heitz, A. (1985) *J. Nat. Prod.* **48**, 778.
6. Nonaka, G. and Nishioka, I. (1977) *Phytochemistry* **16**, 1265.
7. Endo, K., Takahashi, K., Abe, T. and Hikino, H. (1981) *Heterocycles* **16**, 1311.
8. Rosendal Jensen, S., Kjaer, A. and Juhl Nielsen, B. (1973) *Acta Chim. Scand.* **27**, 367.
9. Messana, I., Sperandei, M., Multari, G., Galeffi, C. and Marini-Bettolo, G. B. (1984) *Phytochemistry* **23**, 2617.
10. Abdullahi Hawa, Nyandat, E., Galeffi, C., Messana, I., Nicoletti, M. and Marini-Bettolo, G. B. (1986) *Phytochemistry* **25**, 2821.

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## PHENYLPROPANOID GLUCOSE ESTERS FROM *PRUNUS BUERGERIANA*

HIROKO SHIMOMURA, YUTAKA SASHIDA and TOKUO ADACHI

Tokyo College of Pharmacy, 1432-1, Horinouchi, Hachioji, Tokyo 192-03, Japan

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**Key Word Index**—*Prunus buergeriana*; Rosaceae; phenylpropanoid glucose esters; caffeic acid esters; *p*-coumaric acid esters; cyanogenic glucoside; mandelonitrile glucoside.

**Abstract**—Two new phenylpropanoid glucose esters, 6-*O*-caffeoyl-1-*O*-*p*-coumaroyl- $\beta$ -D-glucopyranose, and 6-*O*-*p*-coumaroyl-D-glucopyranose, along with three known compounds, 1,6-di-*O*-caffeoyl- $\beta$ -D-glucopyranose, 6-*O*-caffeoyl-D-glucopyranose and (2*R*)-[(6-*O*-caffeoyl)- $\beta$ -D-glucopyranosyloxy]benzeneacetonitrile were characterized from the bark of the *Prunus buergeriana* using spectroscopic methods.

## INTRODUCTION

We have previously reported on the isolation and structural determination of a series of phenylpropanoid glucosides from the bark of *Prunus grayana* Maximowicz [1, 2]. In our continuing chemical examination of phenolic compounds in *Prunus* species, we have now isolated two new phenylpropanoid glucose esters from the bark of *Prunus buergeriana* Miquel. This paper describes the isolation and characterization of these compounds.

## RESULTS AND DISCUSSION

A methanolic extract of *P. buergeriana* bark was partitioned with chloroform, and then *n*-butanol. The *n*-butanol soluble part was repeatedly chromatographed

over silica gel and Sephadex LH-20 column to give compounds **1**–**5** as amorphous powders.

Compound **1** was analysed for  $\text{C}_{24}\text{H}_{24}\text{O}_{12}$  (secondary ion mass spectrometry [SIMS]  $m/z$  505 [ $\text{M} + \text{H}$ ] $^+$ ). The  $^1\text{H}$  NMR spectrum of **1** showed the existence of two *trans*-olefin systems, aromatic protons of two ABC systems and sugar protons. An anomeric proton signal ( $\delta$  5.60, *d*,  $J = 7.7$  Hz) indicated that the C-1 position of the glucose moiety was acylated. In the  $^{13}\text{C}$  NMR spectrum, nine pairs of duplicated signals were observed, which were assigned to the phenylpropanoid moieties. On alkaline methanolysis with methanolic sodium methoxide **1** afforded methyl caffeate and D-glucose. Therefore, caffeic acid was attached to some position of 1-*O*-caffeoyl- $\beta$ -D-glucopyranose. The location of the residual caffeoyl group was determined to be the C-6 position of the glucose moiety from the chemical shift value in the  $^{13}\text{C}$  NMR spectrum of